



# Mineral Evolution

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(CLASS)**

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# “You are not in Kansas anymore”

- Robert Hazen and colleagues (2008) had a fundamental insight on mineralogical evolution.
- The mineralogy of terrestrial planets and moons evolves as a consequence of varied physical, chemical, and biological processes that lead to the formation of new mineral species.
- Mineral evolution is a change over time in....
  - The diversity of mineral species
  - The relative abundances of minerals
  - The compositional ranges of minerals
  - The grain sizes and morphologies of minerals



Uraninite UO<sub>2</sub>



Autunite Ca(UO<sub>2</sub>)<sub>2</sub>(PO<sub>4</sub>)<sub>2</sub> x 8-12 H<sub>2</sub>O

# A Few Definitions

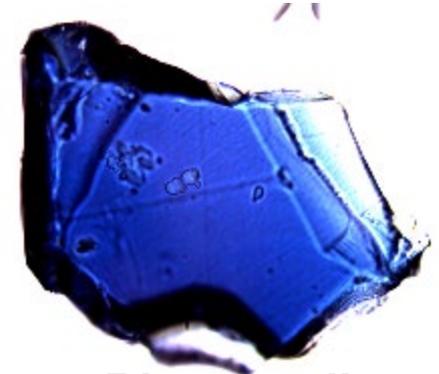
- **Mineral:**
  - A crystalline compound with a fairly well-defined chemical composition and a specific crystal structure.
  - For example, water ice is a mineral.
- **Evolution:**
  - In biology the process by which different kinds of living organisms are thought to have developed and diversified from earlier forms during the history of the earth.
  - More broadly it is the gradual development of something from simple to more complex forms.
- **What we will be talking about is a form of radiation where minerals react in changing chemical and physical environments.**
  - The result are changes to their crystal structure along with their physical and chemical properties.

# Take Olivine

- Olivine is one of the most abundant minerals in the solar system and the universe.
- Forsterite is the Mg-rich endmember:  $\text{Mg}_2\text{SiO}_4$ 
  - Add water and time, it weathers to serpentine  $\text{Mg}_3\text{Si}_2\text{O}_5(\text{OH})_4$
  - Add high pressure the chemistry stays the same, but the crystal structure transforms to Ringwoodite.
  - Add more pressure and it decomposes to silicate perovskite  $\text{MgSiO}_3$  and ferropericlase  $\text{MgO}$
  - In silica-rich igneous systems it reacts to form orthopyroxene  $\text{Mg}_2\text{Si}_2\text{O}_6$
  - Heat olivine under reducing conditions and you get enstatite  $\text{MgSiO}_3$  plus free oxygen and pure Mg.
- So by changing the local chemistry, energy, or pressure, this mineral can “evolve” into 6 more minerals (actually a lot more).



Olivine



Ringwoodite



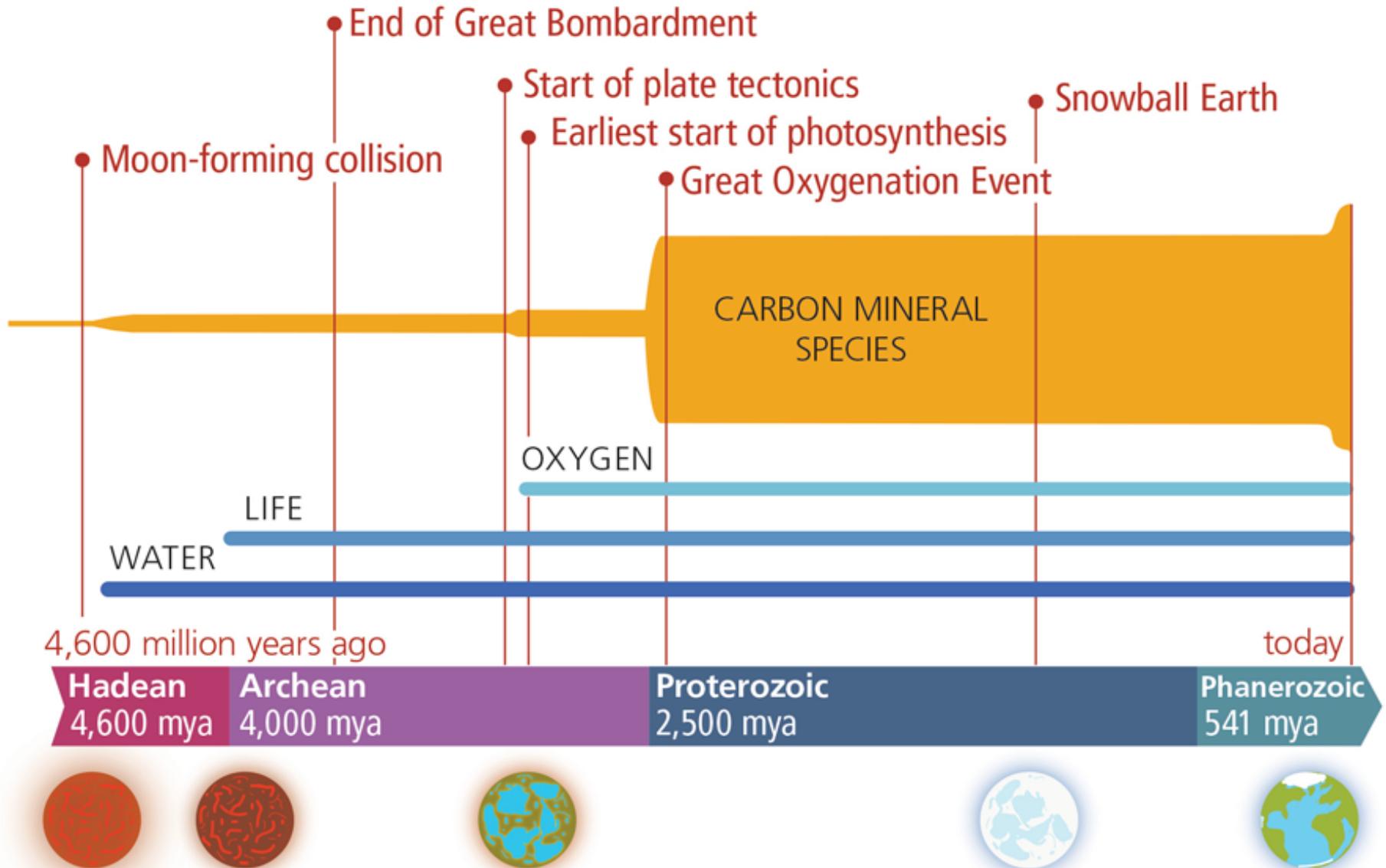
Serpentine

# Mineral Evolution

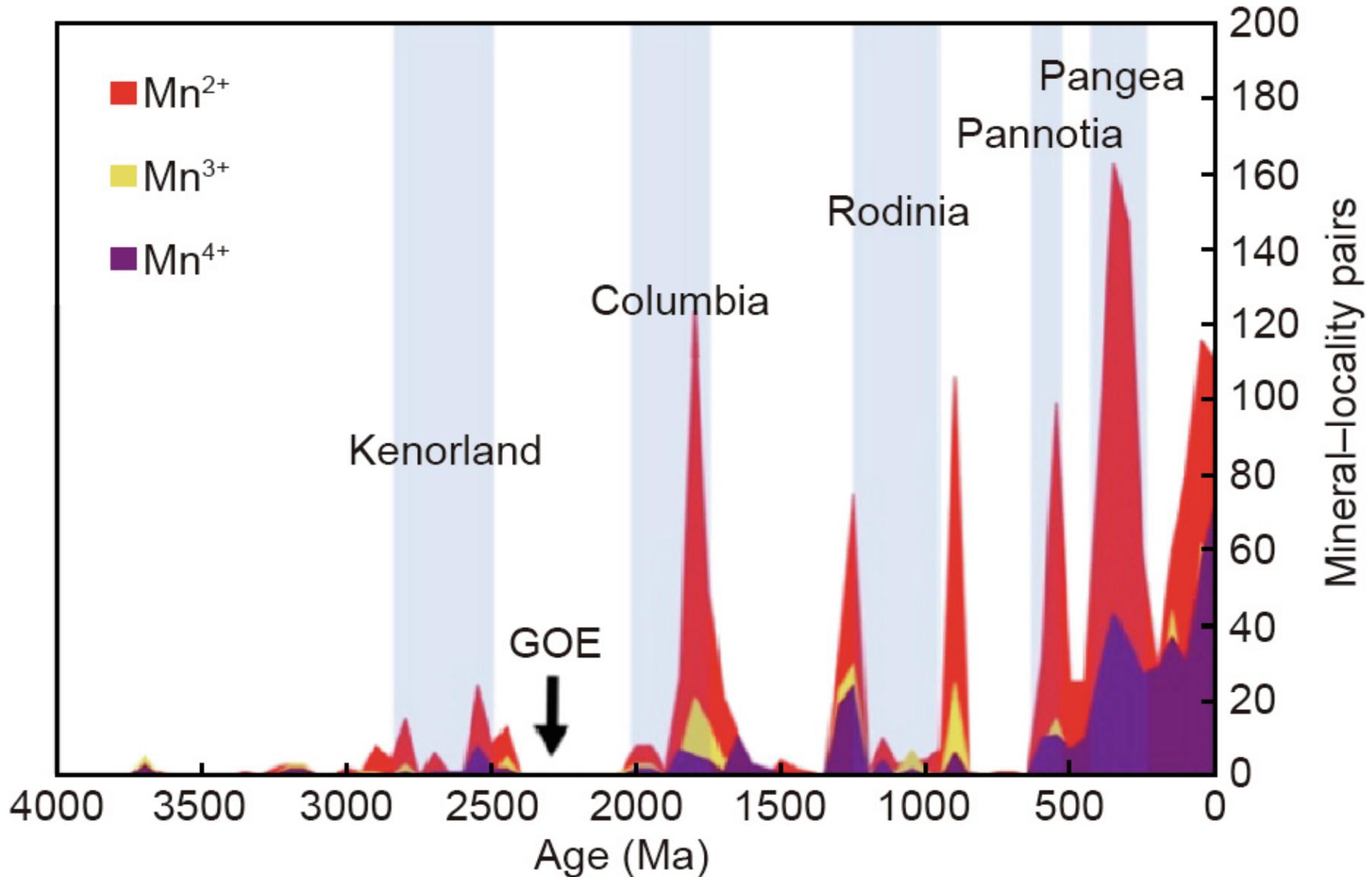
- Mineral inventory of the solar system has gone from about a dozen minerals in the forming solar nebula to over 5400 currently identified on Earth (as of November).
- Three processes drive mineral evolution
  - The progressive separation and concentration of chemical elements from their original uniform distribution.
  - Greater ranges of temperature and pressure coupled with the action of volatiles.
  - The generation of far-from-equilibrium conditions by living systems.
- A few examples of mineral radiation over Earth history.....



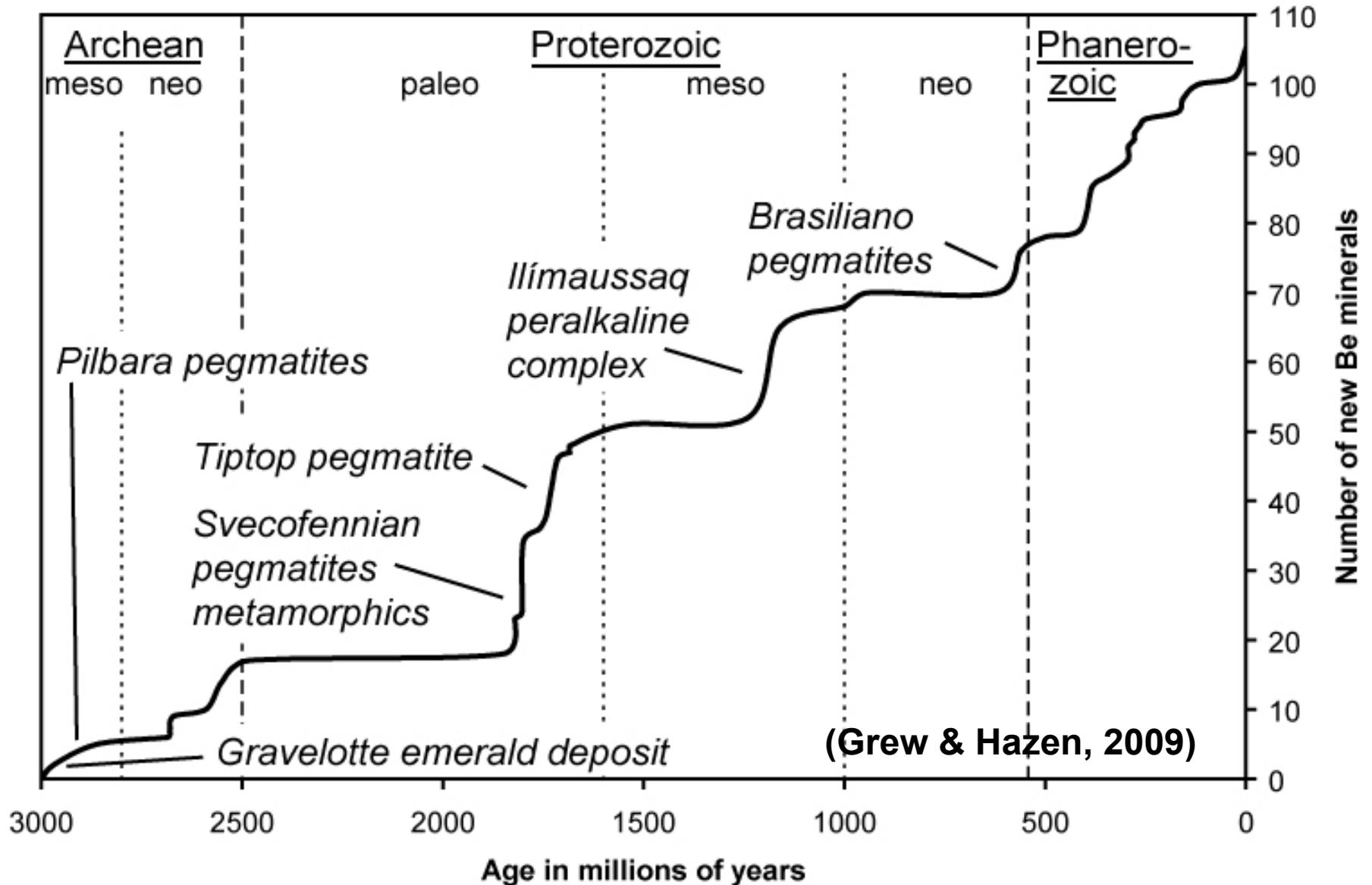
# Diversity of Carbon Minerals



The evolution of manganese minerals over time. Closely associated with the supercontinent cycle and the Earth's near-surface oxidation state. (Hazen et al., 2019)



# No Beryllium Minerals Known Before ~3.0 Gys.



# Stages of Mineral Evolution

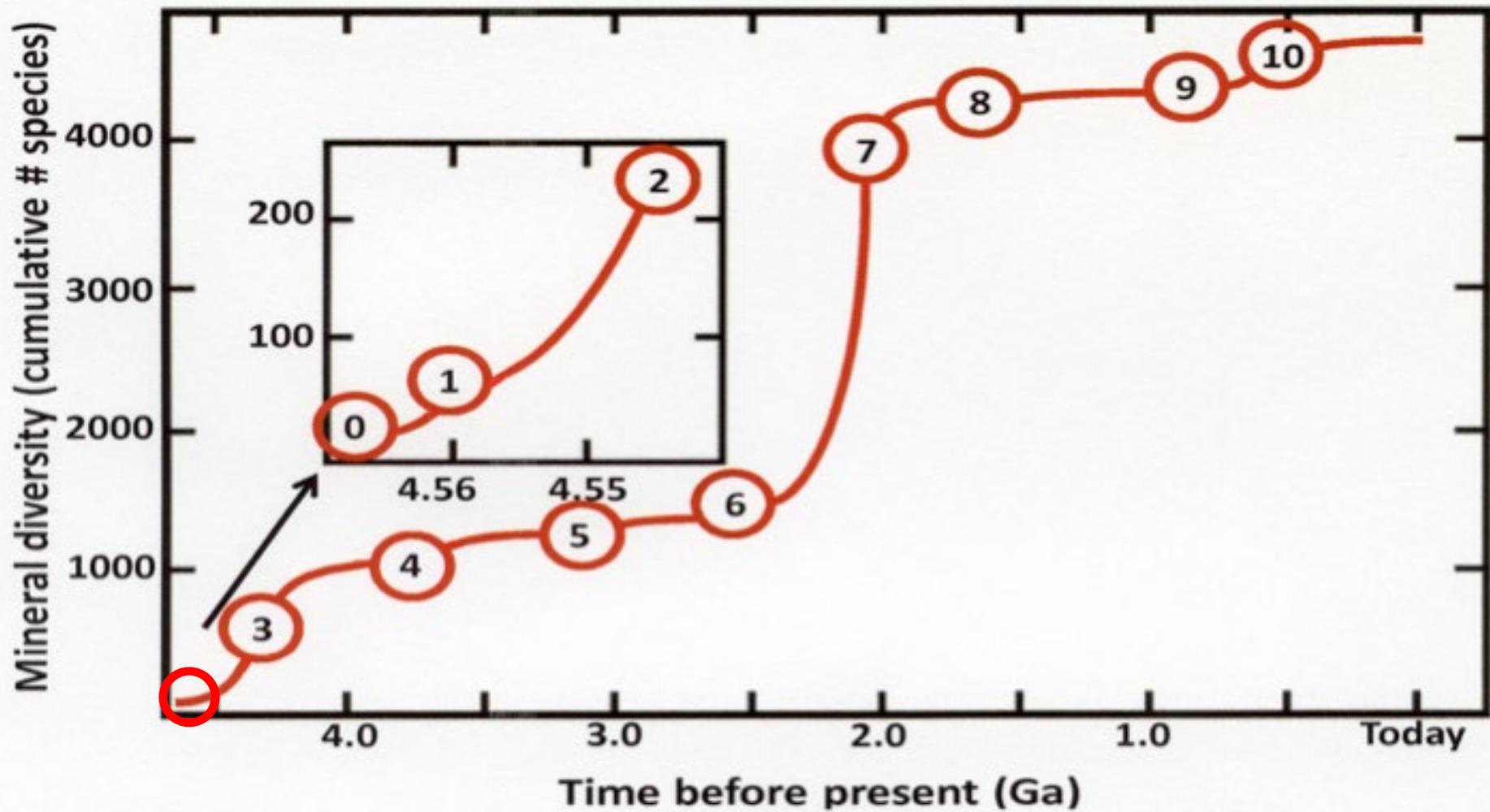
Era/Stage	Age (Ga)	Cumulative no. of species
<b>Prenebular “Ur-Minerals”</b>	>4.6	12
<b>Era of Planetary Accretion (&gt;4.55 Ga)</b>		
1. Primary chondrite minerals	>4.56 Ga	60
2. Achondrite and planetesimal alteration	>4.56 to 4.55 Ga	250
<b>Era of Crust and Mantle Reworking (4.55 to 2.5 Ga)</b>		
3. Igneous rock evolution	4.55 to 4.0 Ga	350 to 500*
4. Granite and pegmatite formation	4.0 to 3.5 Ga	1000
5. Plate tectonics	>3.0 Ga	1500
<b>Era of Biologically Mediated Mineralogy (&gt;2.5 Ga to Present)</b>		
6. Anoxic biological world	3.9 to 2.5 Ga	1500
7. Great Oxidation Event	2.5 to 1.9 Ga	>4000
8. Intermediate ocean	1.9 to 1.0 Ga	>4000
9. Snowball Earth events	1.0 to 0.542 Ga	>4000
10. Phanerozoic era of biomineralization	0.542 Ga to present	4400+

# Stages of Mineral Evolution

		Era/Stage	Age (Ga)	Cumulative no. of species
Planetary		<b>Prenebular “Ur-Minerals”</b>	>4.6	12
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Terrestrial		3. Igneous rock evolution	4.55 to 4.0 Ga	350 to 500*
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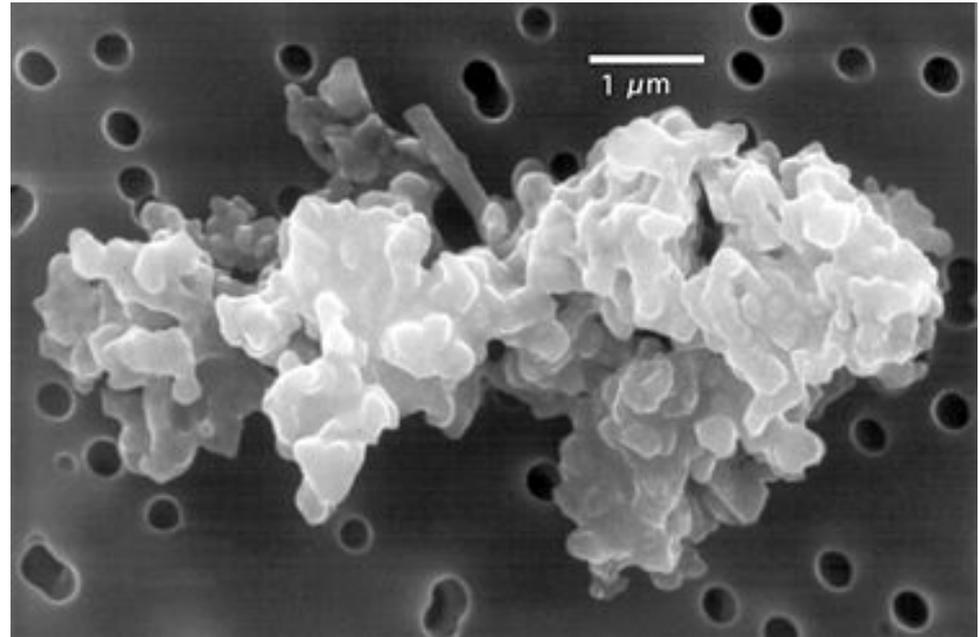
**Asteroid Line** (indicated by a blue dashed line between rows 2 and 3)

**Lunar Line** (indicated by a red dashed line between rows 3 and 4)



# Stage 0: Presolar Grains

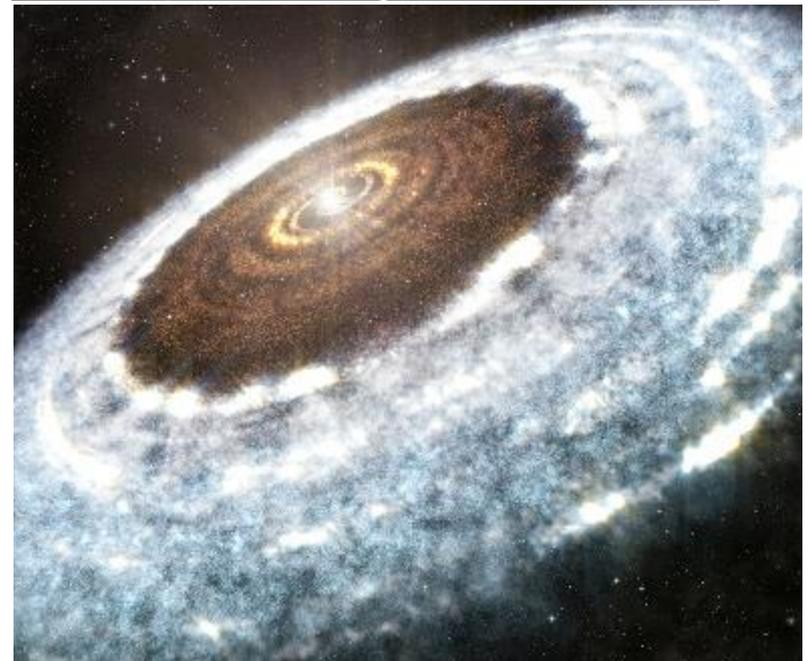
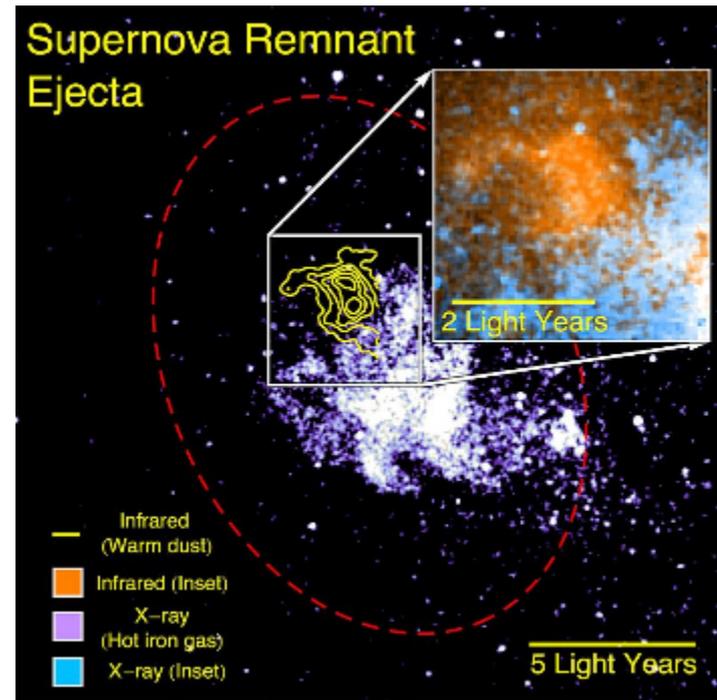
- Presolar stardust grains comprise about 0.1 percent of the total mass of meteorites.
- nitrides
  - Osbornite (TiN)
  - Nierite ( $\alpha$ -Si<sub>3</sub>N<sub>4</sub>);
- carbides
  - Cohenite [(Fe,Ni,Co)3C]
  - Moissanite (SiC)
  - Titanium carbide (TiC)
  - Diamond, graphite (C)
- Iron alloys
  - Kamacite (Fe,Ni)
- oxides
  - Rutile (TiO<sub>2</sub>)
  - Corundum (Al<sub>2</sub>O<sub>3</sub>)
  - Cspinel (MgAl<sub>2</sub>O<sub>4</sub>)
  - Hibonite (CaAl<sub>12</sub>O<sub>19</sub>)
- Silicates
  - Forsterite (Mg<sub>2</sub>SiO<sub>4</sub>)
  - Perovskite-structured MgSiO<sub>3</sub>



- There are probably more minerals to be found.
- Much of the presolar material observed in IDPs and primitive chondrites is amorphous, nonstoichiometric, or partially crystalline.
- Suggests that a much more robust selection of minerals was accreted to form the early solar system.

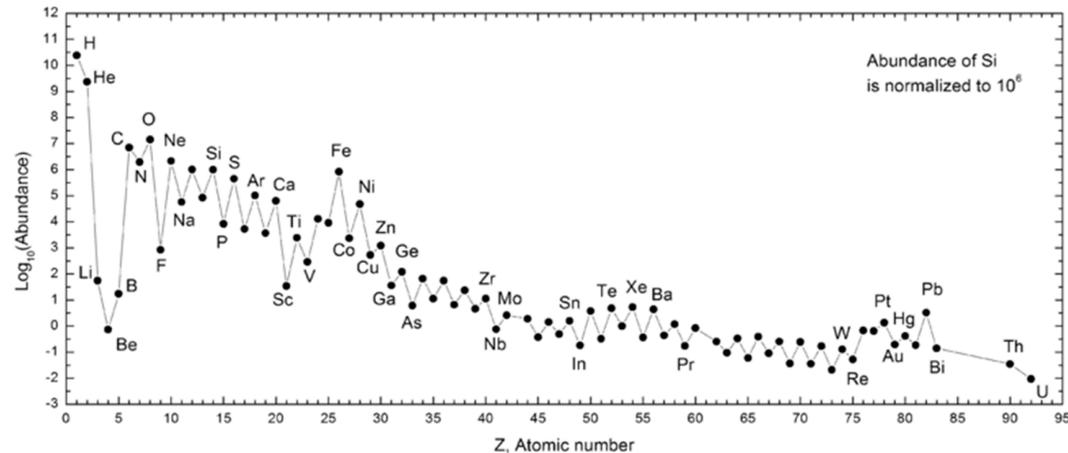
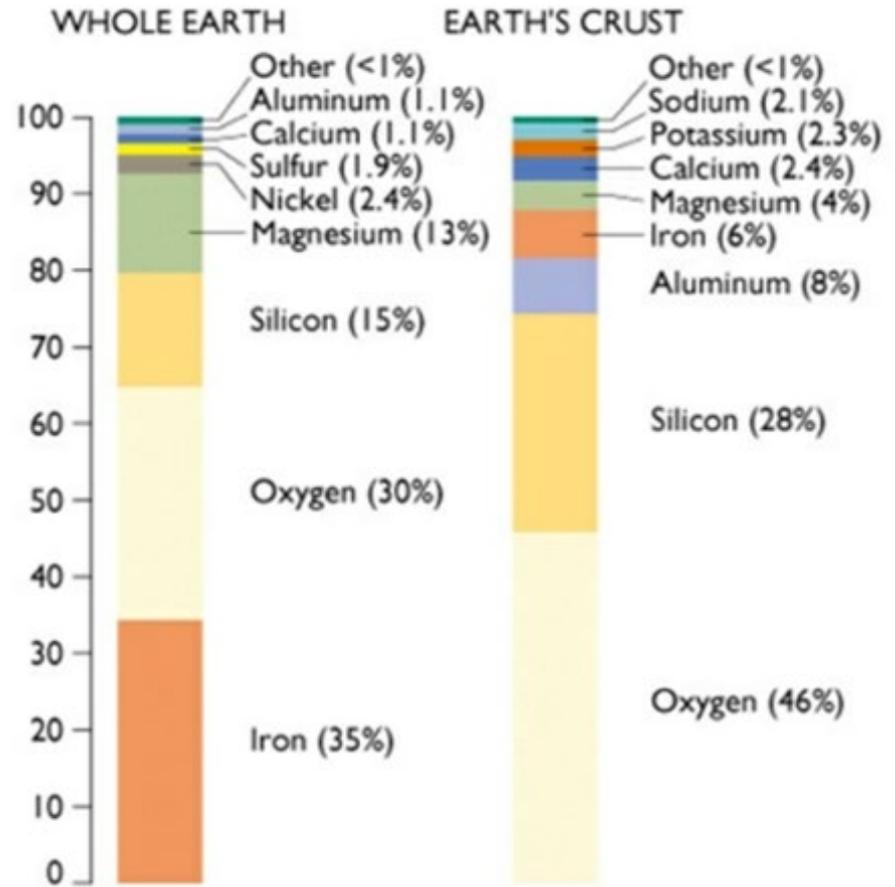
# Solar System Formation

- Solar nebula collapses under self-gravity.
- The collapse heats the nebular material....amount depends on location.
- The protoplanetary nebula is seeded by nearby supernova with materials rich in short-lived radioisotopes.
  - $\text{Al}_{26}$
  - $\text{Fe}_{60}$



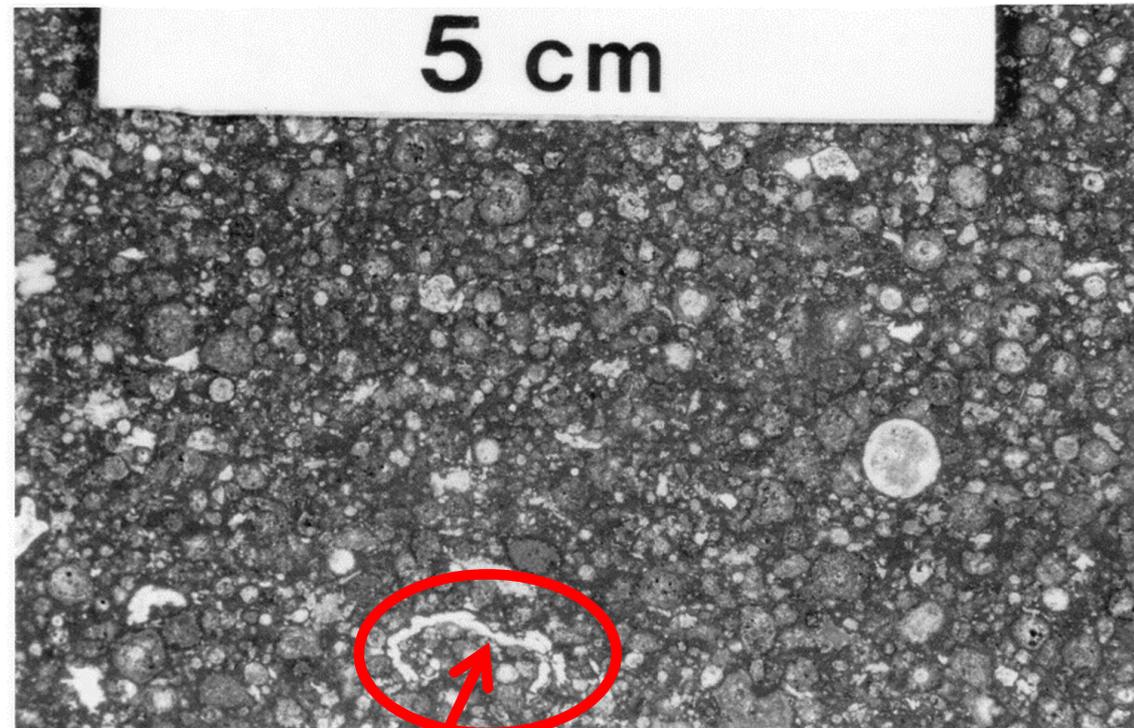
# Elemental Abundances

- What you can build from the nebula depends on what is available.
- What drives initial abundances is the saw-tooth pattern of Nucleosynthesis



# Stage 1: Accretion and the Formation of Chondritic Minerals

- Minerals condense out of the cooling solar nebula, with high-temperature minerals condensing first.
- These were the calcium–aluminum inclusions (CAIs) and include ~24 mineral phases:
  - Spinel ( $\text{MgAl}_2\text{O}_4\text{--FeAl}_2\text{O}_3$ ),
  - Melilite ( $\text{Ca}_2\text{Al}_2\text{SiO}_7$ ) to ( $\text{Ca}_2\text{MgSi}_2\text{O}_7$ )
  - Perovskite ( $\text{CaTiO}_3$ )
  - Hibonite ( $(\text{Ca,Ce})(\text{Al,Ti,Mg})_{12}\text{O}_{19}$ )
  - Calcic pyroxene ( $\text{CaMgSi}_2\text{O}_6$ )
  - Anorthite ( $\text{CaAl}_2\text{Si}_2\text{O}_8$ )
  - Forsterite ( $\text{Mg}_2\text{SiO}_4$ )
- How do we know? ~70,000 recovered meteorites.

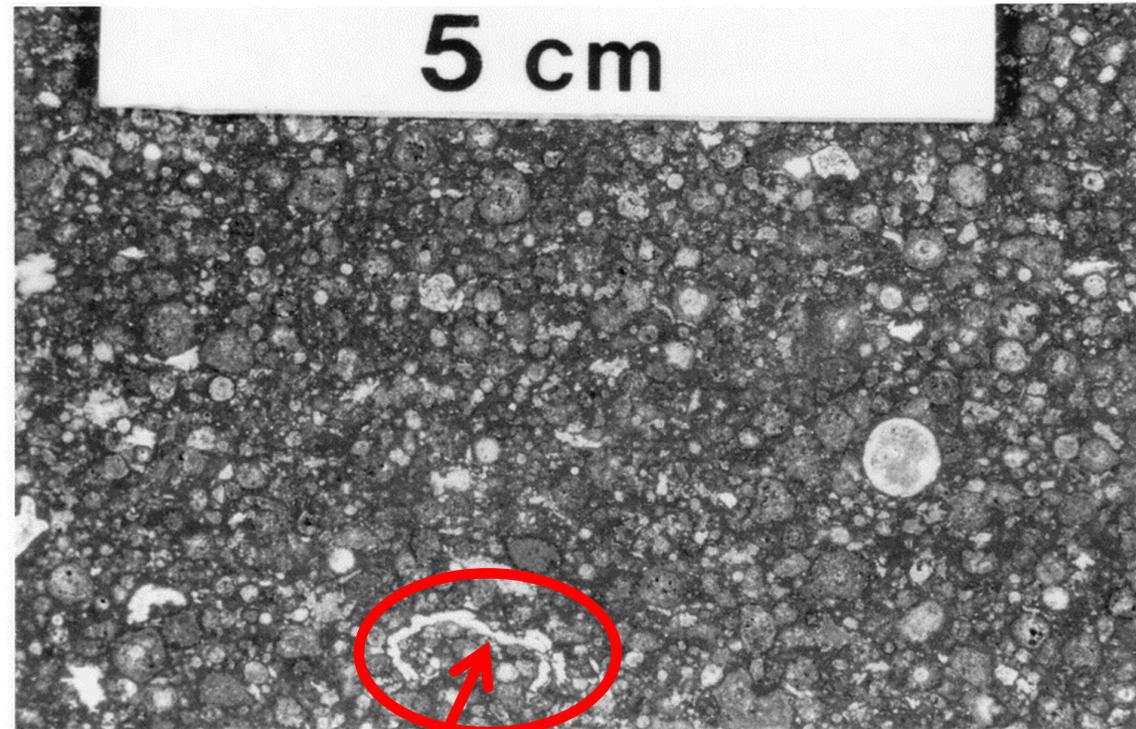


CAI

# Stage 1:

## Cooling and Condensation Continue

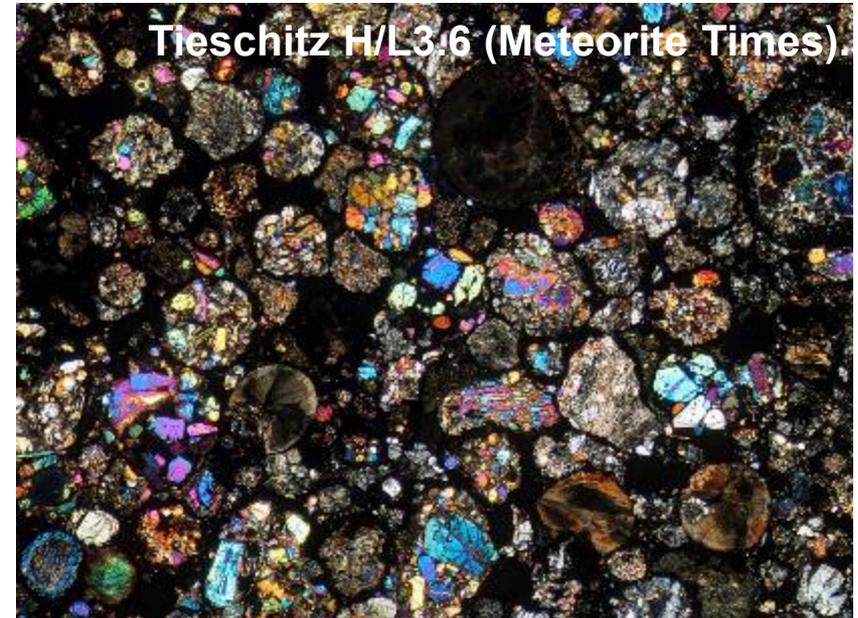
- The cooling nebula condenses progressively lower-temperature minerals.
- Chondrules dominate this stage.
- Chondrules are the sedimentary “sand” of the solar system....which make up the chondrite meteorites.
  - Millimeter sized spheres formed during flash melting in the solar nebula.
- Chondrites contain a diversity of metals, sulfides, oxides, and phosphates.



CAI

# Stage 1: Primary Chondritic Minerals

- The mineral assemblage at this stage is ~60 minerals
- Formed under chemically diverse environments, particularly oxygen fugacity which ranges from highly reduced enstatite chondrites to the oxidized carbonaceous chondrites.
- Dominant minerals include:
  - Olivine
  - Pyroxene
  - Plagioclase
  - FeNi
  - Troilite
- Metamorphism, aqueous alteration, and shock will alter, modify, and diversify the chondrites.



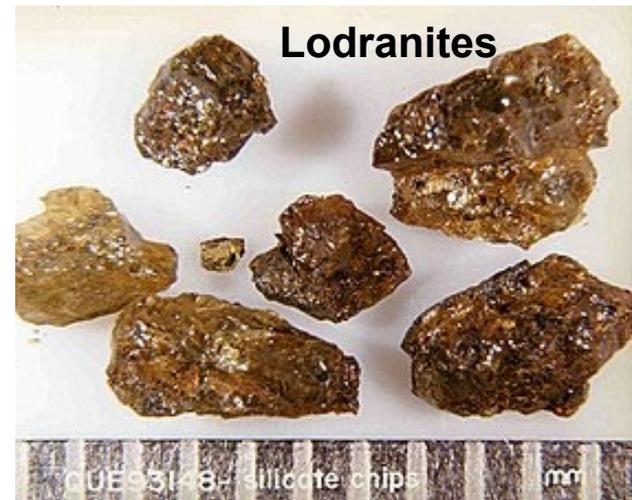
# Stage 2

## Achondrite and Planetesimal Alteration

- Chondrules accrete into planetesimals, planetesimals accrete into planets over about 10 million years.
- The timing planetesimal growth depends on location and the local density of materials.
- Heating becomes a critical factor.
- Heat sources:
  - Gravitational potential energy from accretion
  - Radioactive isotopes
  - Core formation
- Remember that the nebula was seeded with material from a nearby supernova.
  - Included in the seeding was very short-lived nuclides including  $^{26}\text{Al}$  (717,000 years) and  $^{60}\text{Fe}$  (2.6 million years)
  - Because of the abundance and short half-life of  $^{26}\text{Al}$ , its heat generation potential is about 1,000,000 times that of Uranium.



NWA 4231 (ureilite)



Lodranites

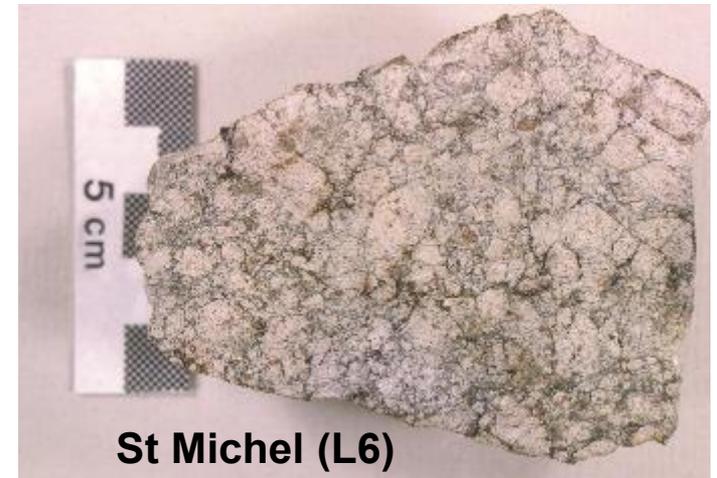
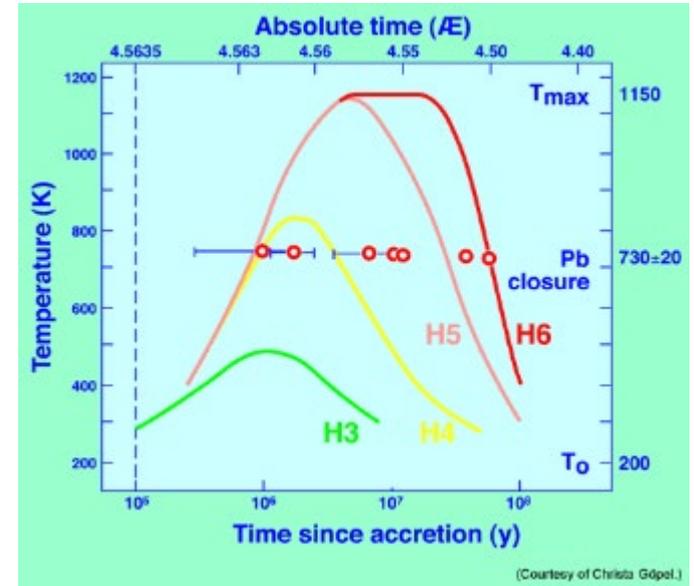
CU E 31 48- silicate chips

mm

# Stage 2

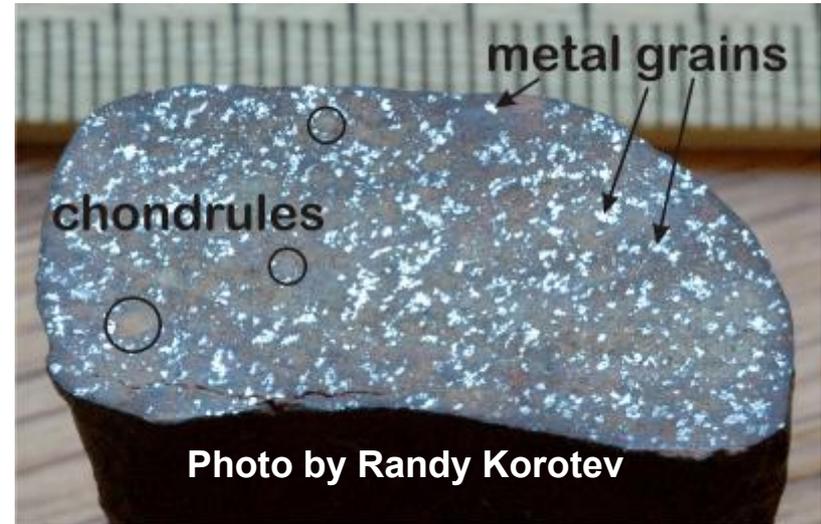
## Timing and Size are Key to Outcomes

- Planetesimals that accrete early (with lots of  $^{26}\text{Al}$ ) heat to melting and differentiate, producing igneous melts.
- With the short half-life of  $^{26}\text{Al}$  it does not take much time before the isotope is depleted and the heating will only metamorphize the planetesimal.
- Dr. Steve Desch will talk about planetesimal accretion and timing in detail.
- Planetesimals that accrete outside the “frost line” will include frozen volatiles as well as minerals. Heating of this assemblage will produce aqueous alteration.
- Planetesimals that accrete outside the frost line and later do not heat much, producing comets that retain unaltered refractory minerals and ices.



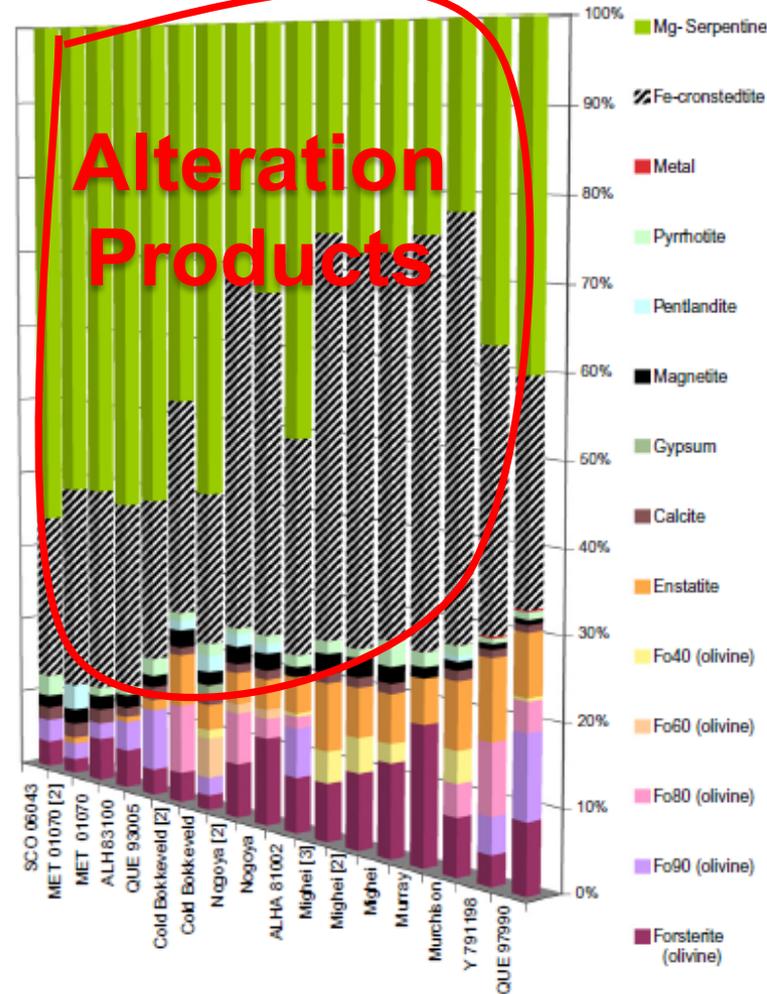
## Stage 2: Metamorphism

- Heating of anhydrous ordinary and enstatite chondrites produced new minerals from thermal metamorphism at temperatures up to  $\sim 950^{\circ}\text{C}$
- **Phosphates**
  - Apatite  $\text{Ca}_5(\text{PO}_4)_3(\text{F},\text{Cl},\text{OH})$
  - Merrillite  $\text{Ca}_9\text{NaMg}(\text{PO}_4)_7$
- **Silicates**
  - Nepheline  $(\text{Na},\text{K})\text{AlSiO}_4$
- **Oxides**
  - Rutile  $\text{TiO}_2$
  - Quartz  $\text{SiO}_2$  and its high-temperature polymorphs Cristobalite and Tridymite



# Stage 2: Aqueous Alteration

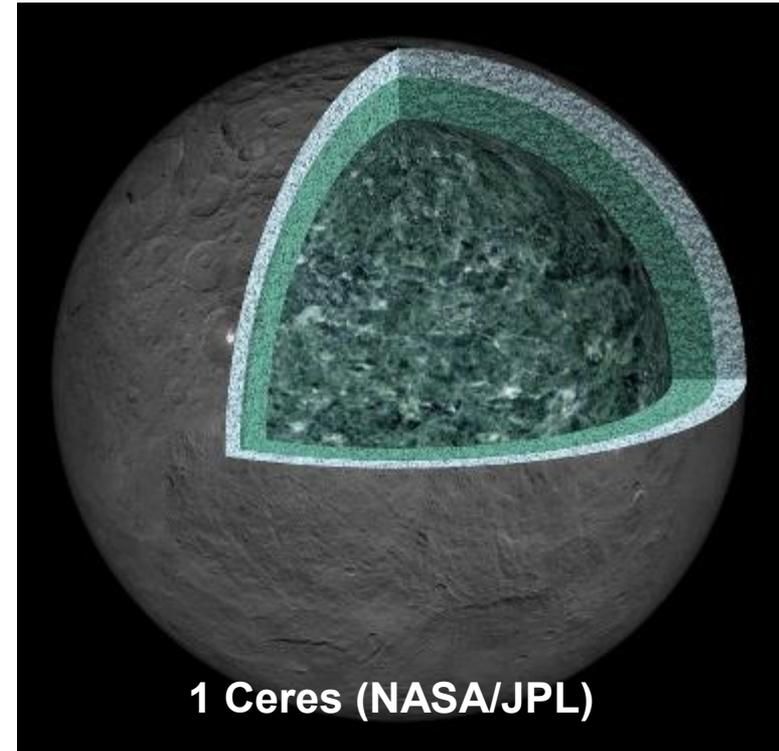
- The melting of ice and the subsequent alteration of chondrule silicates at low temperatures (<100°C) produced a range of new minerals
- **Phyllosilicates**
  - Montmorillonite  
( $\text{Na}, \text{Ca}_{0.5}$ )<sub>0.33</sub>(Al, Mg)<sub>2</sub>Si<sub>4</sub>O<sub>10</sub>(OH)<sub>2</sub>·nH<sub>2</sub>O
  - Chrysotile Mg<sub>3</sub>Si<sub>2</sub>O<sub>5</sub>(OH)<sub>4</sub>
  - Cronstedtite (Fe<sub>2</sub><sup>2+</sup>, Fe<sup>3+</sup>)<sub>3</sub>(Si, Fe<sup>3+</sup>)<sub>2</sub>O<sub>5</sub>(OH)<sub>4</sub>
- **Oxides**
  - Magnetite Fe<sup>2+</sup>Fe<sup>3+</sup><sub>2</sub>O<sub>4</sub>
  - Ferrihydrite 5Fe<sub>2</sub>O<sub>3</sub>·9H<sub>2</sub>O
- **Sulfides**
  - Pyrrhotite Fe<sub>1-x</sub>S (x = 0 to 0.2)
  - Pentlandite (Fe, Ni)<sub>9</sub>S<sub>8</sub>
- **Carbonates**
  - Dolomite CaMg(CO<sub>3</sub>)<sub>2</sub>
  - Calcite CaCO<sub>3</sub>
- **Sulfates**
  - Gypsum CaSO<sub>4</sub>·2H<sub>2</sub>O
  - Epsomite MgSO<sub>4</sub>·7H<sub>2</sub>O



From Howard et al, 2011

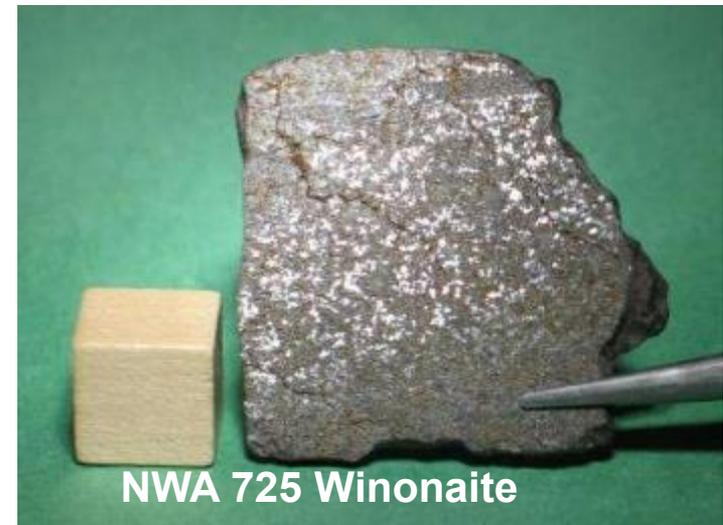
# Stage 2: Aqueous Alteration

- On small near-Earth asteroids the only surviving volatiles are in the mineral alteration products (hydrated phyllosilicates, sulfates, carbonates, oxides).
- Aqueous processes on most asteroids are short-lived and open system. Water not incorporated into minerals either migrated to the surface and sublimated or was retained as ice in the subsurface.
- The largest main-belt volatile-rich asteroids may have a mantle of hydrated rocks with a crust of ice, salts, and hydrated minerals. Salt-rich brines may be occasionally active. Ceres for example, the mantle is estimated to be 23-28 wt.% water.



# Stage 2: Igneous Alteration

- In some planetesimals heating continued above  $\sim 950^{\circ}\text{C}$  crossing the liquidus for early partial melts from FeNi metal and troilite.
- The early melts migrated through the unmelted silicates and form meteorites like acapulcoites, winonaites, and IAB irons.
- As the temperatures increased, silicate melting formed pyroxene–plagioclase-rich melts.
  - Residual rocks remaining after silicate melting are represented by meteorite groups like the ureilites and lodranites.
- The partial melts sequestered a range of incompatible elements, including phosphorus, sulfur, and carbon which reacted with unmelted silicates to form new minerals.
  - Phosphates: Na–Ca–Mg phosphates chladniite, panethite, brianite, and johnsomervilleite
  - Carbides: cohenite and haxonite  $(\text{Fe,Ni})_{23}\text{C}_6$ .



Group →	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
↓ Period																		
1	1 H																	2 He
2	3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
3	11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
4	19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
5	37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
6	55 Cs	56 Ba	*	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
7	87 Fr	88 Ra	**	(104) Rf	(105) Db	(106) Sg	(107) Bh	(108) Hs	(109) Mt	(110) Ds	(111) Rg	(112) Cn	(113) Nh	(114) Fl	(115) Uup	(116) Lv	(117) Uus	(118) Uuo
			* Lanthanides	57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
			** Actinides	89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	(95) Am	(96) Cm	(97) Bk	(98) Cf	(99) Es	(100) Fm	(101) Md	(102) No	(103) Lr

# Elements concentrated in asteroid cores

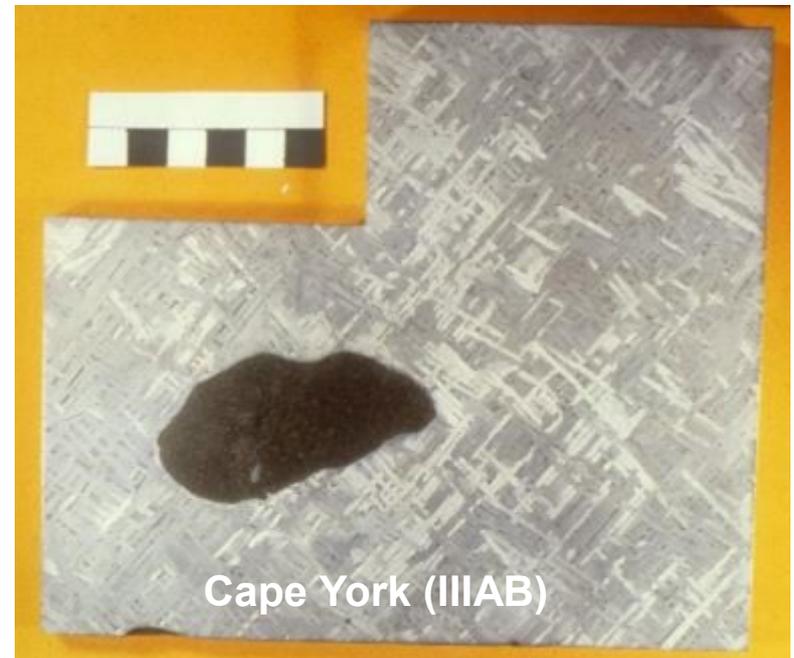
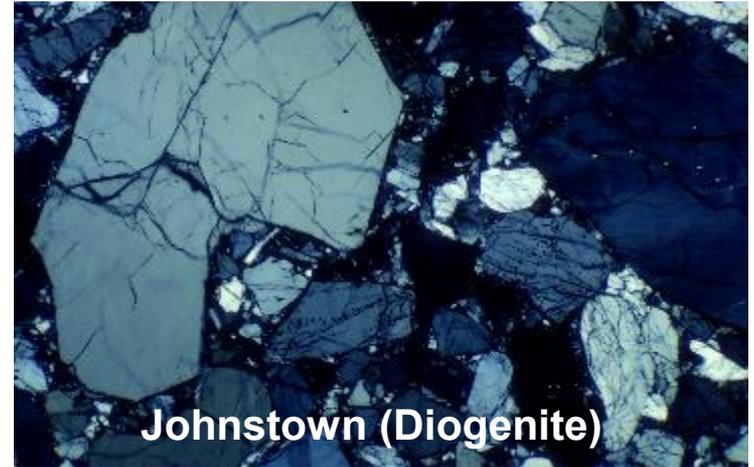
## Legend

Lithophile
  Siderophile
  Chalcophile
  Atmosphile
  very rare

- **Lithophile** (rock loving) elements remain on or close to the surface because they combine readily with oxygen, forming compounds that do not sink into the core.
- **Siderophile** (iron loving) elements are the high-density transition metals which tend to sink into the core because they dissolve readily in iron.
- **Chalcophile** (ore loving) elements that combine readily with sulfur and/or some other chalcogen other than oxygen.
- **Atmosphile** (atmosphere loving) elements are either gases or form volatile hydrides.

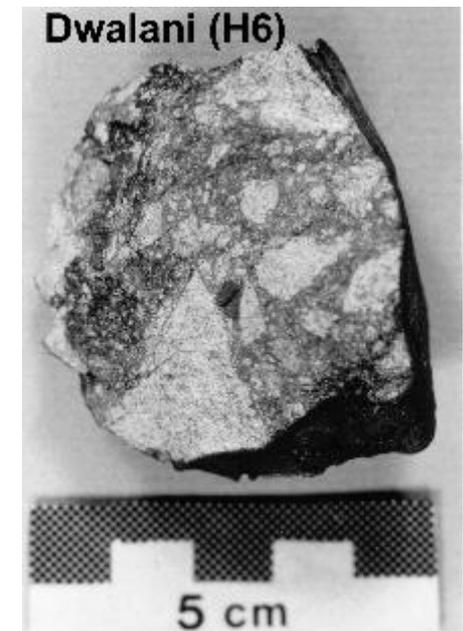
## Stage 2: Igneous Alteration

- At high degrees of melting differentiation sequestered siderophile from lithophile elements that crystallized separately to form the crust, mantle, and core of the planetesimal.
- Within the crust more incompatible elements were concentrated.
  - Feldspar ( $\text{KAlSi}_3\text{O}_8$ ), titanite ( $\text{CaTiSiO}_5$ ), zircon ( $\text{ZrSiO}_4$ ), and baddeleyite ( $\text{ZrO}_2$ ) formed.
- In the core, mineralogical diversity was controlled both by fractional crystallization and solid-state transformations during cooling.



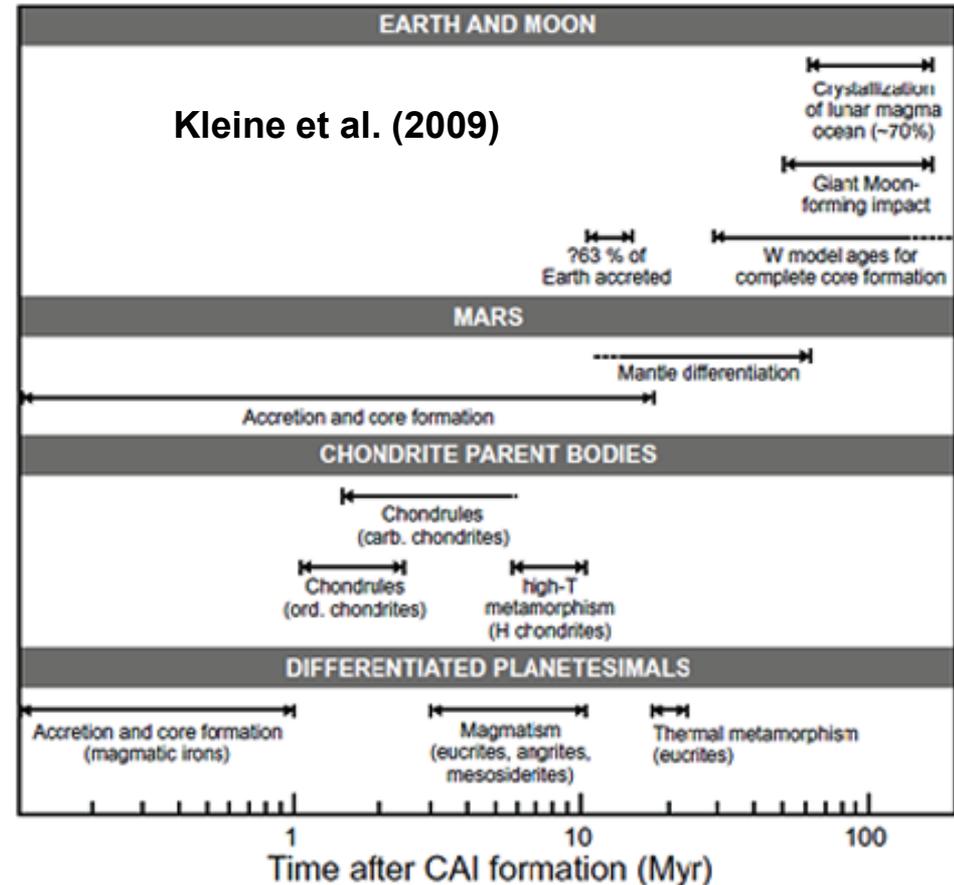
# Impacts and Mineral Evolution

- **Formation of shock minerals.**
  - High-pressure minerals in meteorites are often polymorphs of lower pressure common minerals
  - Olivine → ringwoodite
  - Chondrite melt → majorite garnet
  - Magnesiowüstite
  - Enstatite → akimotoite
  - Plagioclase feldspar → maskelynite
- **For asteroids impact fracture and rubblize the bodies. Asteroids in near-Earth space are collisional fragments and probably rubble piles.**
- **On larger bodies, excavation of deep igneous and metamorphic terrains can initiate hydrothermal activity.**
- **Creation of deep subsurface hydrothermal zones (if there is water.....i.e. Mars).**



# The End of Asteroid Mineral Evolution

- In asteroids heat drives mineral evolution.
- But heat from accretion, core formation, and strong radioactive sources was exhausted early in solar system history.
- Metamorphism lasted longest on large asteroids, but ended within 20-30 Myr.
- After this period, the only mineralization action was impact-related.
- Between original mineralogy, aqueous, metamorphic, impact, and igneous evolution meteorites have about 250 minerals.



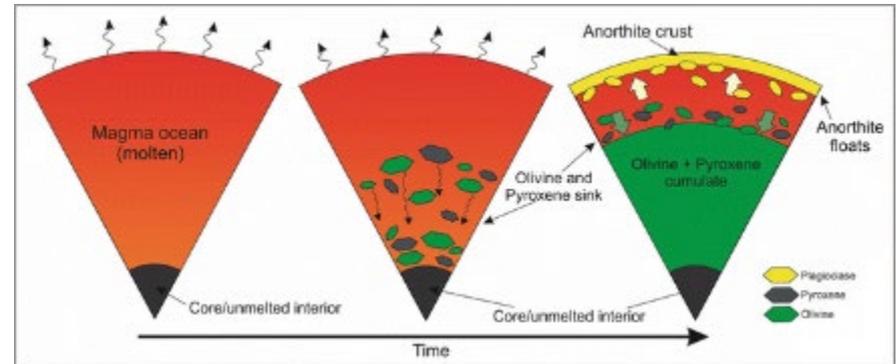
# Stage 3: The Moon

- The moon was the product of a giant impact that ripped off the crust and some of mantle of the early Earth.
- The giant impact was between two already differentiated planets.
- It was crust and mantle material that largely formed the Moon.
- That means the starting material was already depleted in siderophile elements.
- Volatiles would have been vaporized by the high impact temperatures.
- Accretional energy liberated during reaccretion would generate a magma ocean.

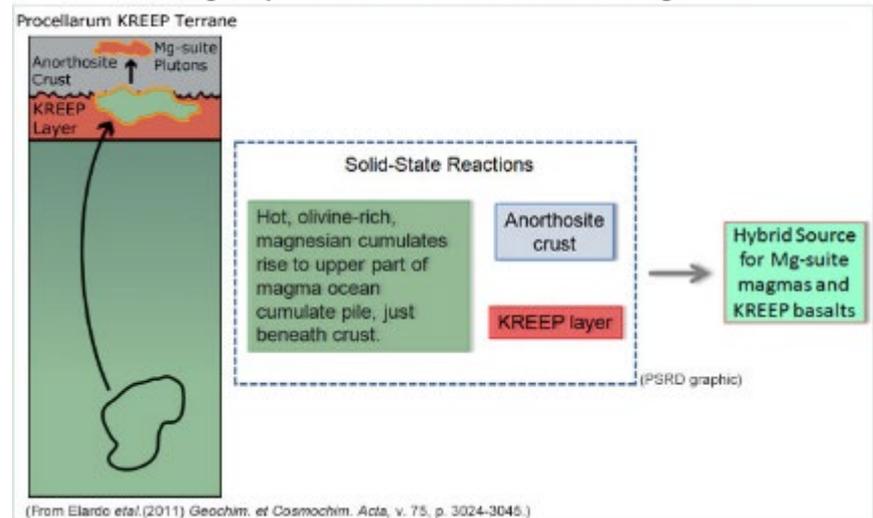


# Stage 3: The Moon

- Hot accretion meant that most of the Moon was initially molten.
- The magma ocean slowly cooled and differentiated. Tidal dissipation provided the energy to slow cooling.
- The ocean differentiated with denser olivine and pyroxene dropping to the bottom and less dense anorthite floating.
- Like igneous asteroids, high degrees of crystallization concentrated incompatible elements in the crust.
  - The KREEP (K for potassium, REE for rare-earth elements and P for phosphorus) source region in Procellarum was formed.
- Also like igneous asteroids mineralogical diversity comes from both by fractional crystallization and solid-state transformations during cooling.

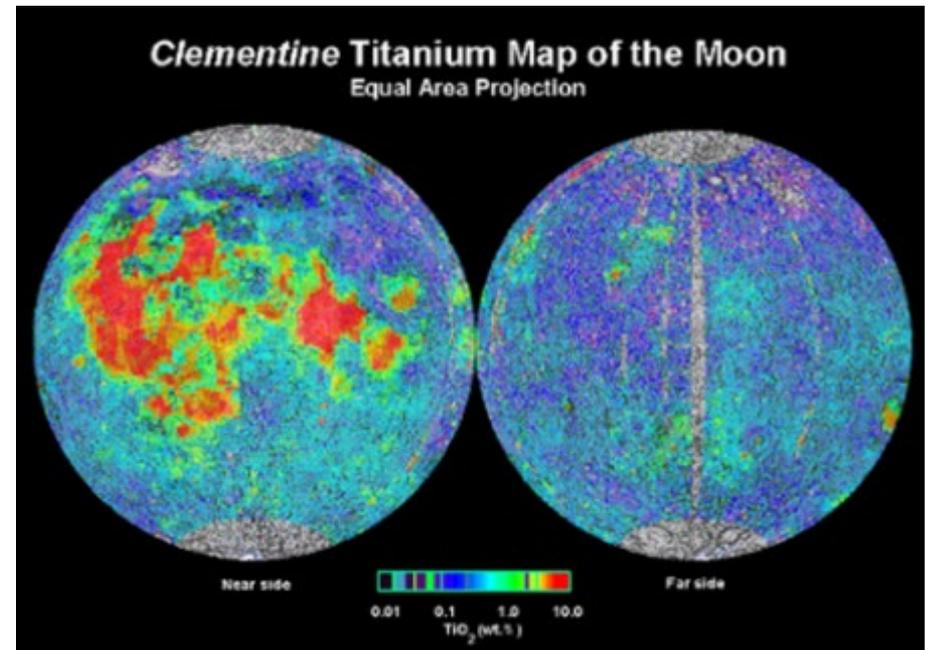
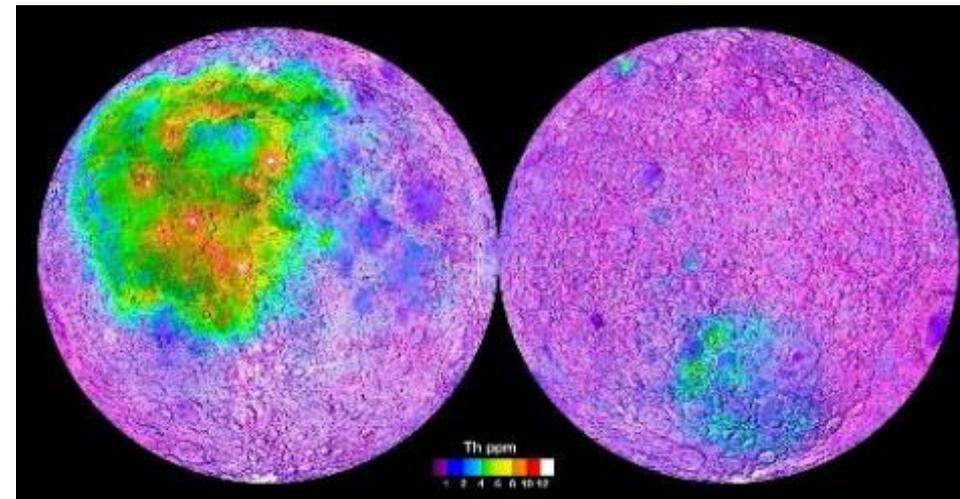


*Rising Diapir Model and Formation of the Mg-Suite*



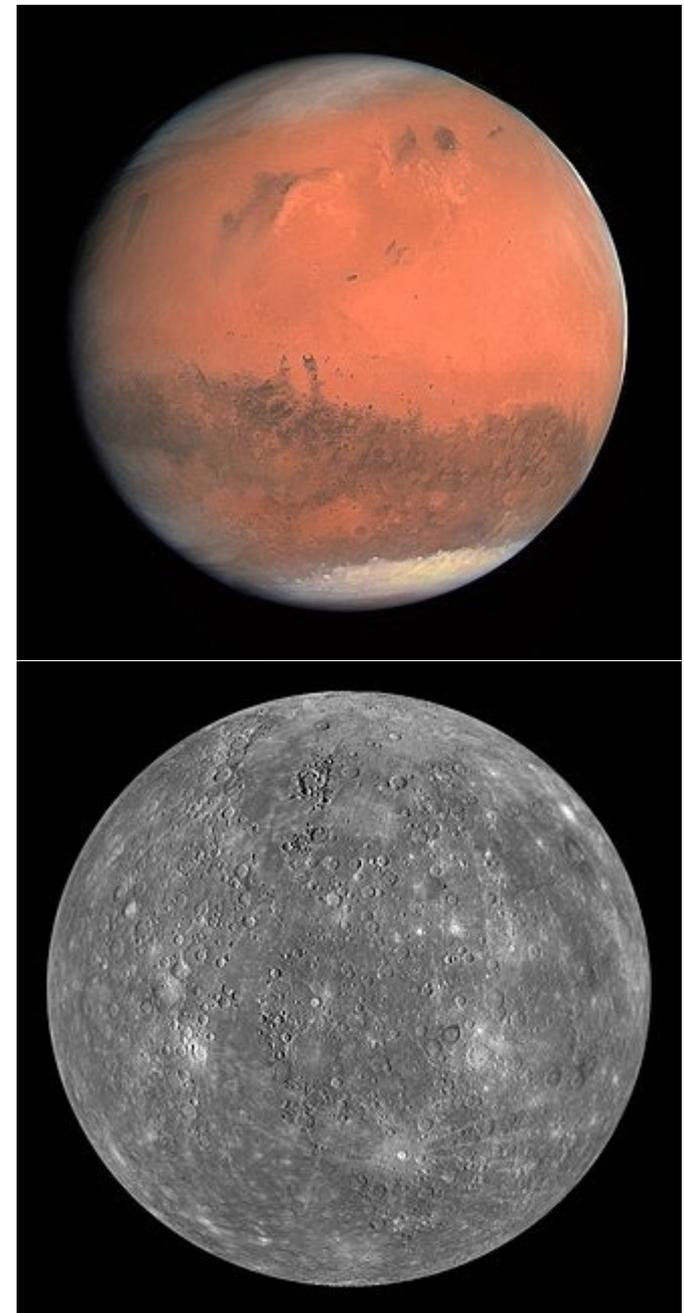
# Stage 3: The Moon

- The KREEP terrain is concentrated in Procellarum and Imbrium.
- A large proportion of the Moon's inventory of heat producing elements was incorporated into the KREEP.
- Mare volcanism between 4.2-3.16 Ga produced the Mare terrains seen on the Lunar near side.
- These basalts tapped the KREEP as part of their source region and material.
- A major difference between terrestrial and Lunar basalts is the near-total absence of water in the lunar basalts. As a result they erupt much hotter and more fluid than terrestrial basalts.



# Stage 3: Planets

- All rocky planets and moons experience Stage 3 mineral forming igneous processes
- Even on a volatile-poor body like Mercury or the Moon, such processes yield as many as 350 different mineral species.
- If water and other volatiles are abundant, then the mineralogical diversity is enhanced by the development of hydroxides, hydrates, carbonates, and evaporite minerals—a total of approximately 500 mineral species.
- A once-wet Mars appears to have progressed this far in its mineral evolution.



# Stages of Mineral Evolution

		<b>Era/Stage</b>	<b>Age (Ga)</b>	<b>Cumulative no. of species</b>
<b>Planetary</b>		<b>Prenebular “Ur-Minerals”</b>	>4.6	12
		<b>Era of Planetary Accretion (&gt;4.55 Ga)</b>		
		1. Primary chondrite minerals	>4.56 Ga	60
		2. Achondrite and planetesimal alteration	>4.56 to 4.55 Ga	250
		<b>Era of Crust and Mantle Reworking (4.55 to 2.5 Ga)</b>		
<b>Terrestrial</b>		3. Igneous rock evolution	4.55 to 4.0 Ga	350 to 500*
		4. Granite and pegmatite formation	4.0 to 3.5 Ga	1000
		5. Plate tectonics	>3.0 Ga	1500
		<b>Era of Biologically Mediated Mineralogy (&gt;2.5 Ga to Present)</b>		
		6. Anoxic biological world	3.9 to 2.5 Ga	1500
		7. Great Oxidation Event	2.5 to 1.9 Ga	>4000
	8. Intermediate ocean	1.9 to 1.0 Ga	>4000	
	9. Snowball Earth events	1.0 to 0.542 Ga	>4000	
	10. Phanerozoic era of biomineralization	0.542 Ga to present	4400+	

**Asteroid Line** (indicated by a blue dashed line between rows 2 and 3)

**Lunar Line** (indicated by a red dashed line between rows 3 and 4)

# Mineral Evolution: Stage 4 and 5

(All we do not see in Lunar and asteroid geology)

- **Stage 4: A planet has enough heat to remelt its initial basaltic crust**
  - Forms granitoids from fractionation of the basalt.
  - Repeated partial melting and concentration of rare elements form pegmatites
  - Approximately 500 distinctive minerals of Li, Be, B, Nb, Ta, U, and a dozen other rare elements
- **Stage 5: Onset of plate tectonics**
  - Subduction of H<sub>2</sub>O-rich crustal materials led to fluid-rock interactions and rare element concentration.
  - Uplift and erosion exposed new high-pressure, low-temperature minerals formed in subduction zones.
  - 500 more minerals.



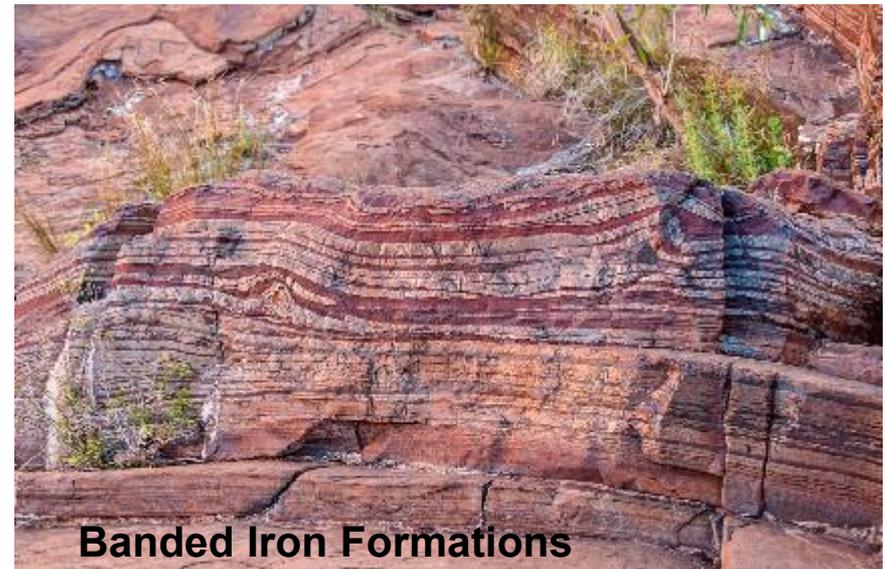
# Mineral Evolution: Stage 6 and 7

(All we do not see in Lunar and asteroid geology)

- **Stage 6: The anoxic biosphere interacts with the lithosphere**
  - Primitive microbes in the anoxic Archean Earth played a relatively minor role in mineralogy.
  - The 1500 mineral species would probably occur in any volatile-rich anoxic terrestrial planet.
- **Stage 7: Great Oxidation Event**
  - The rise of atmospheric oxygen paved the way for more than 2,500 new minerals.
  - Many were hydrated, oxidized weathering products of other minerals.
  - The planet rusted. Black basalt that turned red as the ferrous iron ( $\text{Fe}^{2+}$ ) oxidized to hematite.



Turquoise  $\text{CuAl}_6(\text{PO}_4)_4(\text{OH})_8 \cdot 4\text{H}_2\text{O}$

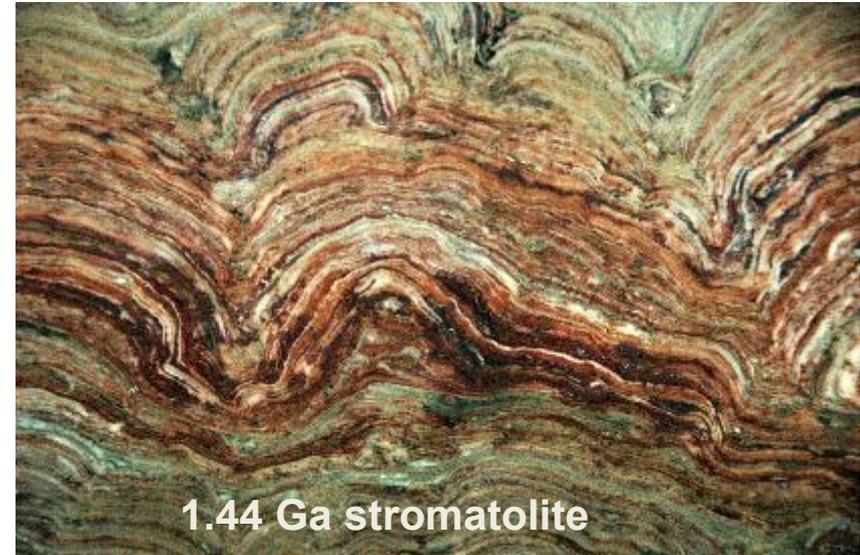


Banded Iron Formations

# Mineral Evolution: Stage 8 and 9

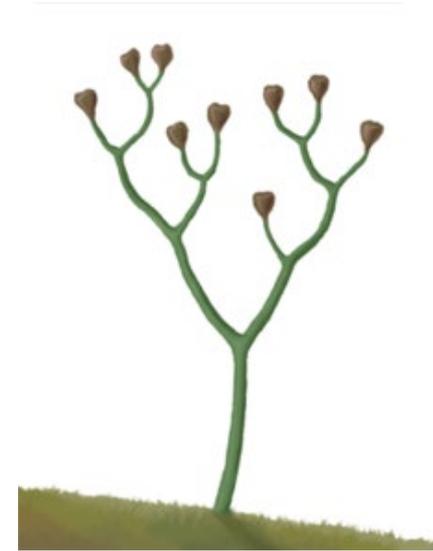
(All we do not see in Lunar and asteroid geology)

- **Stage 8: The intermediate ocean**
  - The “boring billion”.
  - BIF ceased because the ocean chemistry reached an intermediate oxidation state.
  - Minimal mineralogical innovation.
- **Stage 9: Snowball Earth**
  - Fluctuations in climate and atmospheric chemistry produced at least two snowball Earth events.
  - During glaciation surface weathering slowed down allowing for volcanic  $\text{CO}_2$  buildup and then rapid greenhouse deglaciation.
  - Glaciation enhanced weathering of sulfides and the production of clays.



# Mineral Evolution: Stage 10

- **Creation of minerals by living organisms becomes widespread**
- **The buildup of atmospheric oxygen allowed the development of the stratospheric ozone layer, which shielding the surface from solar UV and allowed the start of a terrestrial biosphere.**
- **Allowed for rapid biochemical breakdown of rock.**
- **Increasing weathering rates of basalt, granite and limestone by an order of magnitude.**
- **The abundance of clay minerals and the rate of formation of soils increased vastly.**



**Cooksonia, the earliest vascular plant**





# To Wrap Up



- **Mineral evolution are fundamentally different on the Moon and Asteroids vs. the Earth.**
- **Asteroids have about 250 minerals.**
- **The Moon has about 350 minerals.**
- **Geological concentration mechanisms that we depend on terrestrially for ores do not exist on the Moon and asteroids.**
- **On asteroids no high heat flow, no available fluids, no hydrothermal systems for the last ~4.5 billion years.**
- **On the Moon, no fluids or hydrothermal systems. What you get is Mare volcanism tapping KREEP source regions.**
- **Impacts and shock are the major drivers for most of solar system history.**

